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RESEARCH ARTICLE

BIFUNCTIONALIZED ALLENES. PART XVII. SYNTHESIS OF 2,5-DIHYDRO-1,2-OXAPHOSPHOLES AND 4-PHOSPHORYL-3,6-DIHYDRO-2H-PYRANS BY ELECTROPHILIC CYCLIZATION AND COINAGE METAL-CATALYZED CYCLOISOMERIZATION OF PHOSPHORYLATED β -HYDROXYALLENES

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ABSTRACT

A convenient and efficient regioselective synthesis of phosphorylated β -hydroxyallenes by an atom economical [2,3]-sigmatropic rearrangement of the mediated propargyl phosphite or phosphinite which can be readily prepared *via* reaction of protected 5-methyl-dec-6-yn-5-ol with dimethyl chlorophosphite or chlorodiphenyl phosphine respectively is described. This paper discusses a reaction of phosphorylated β -hydroxyallenes with protected or unprotected hydroxy group involving 5-*endo-trig* cyclization. Reaction of dimethyl [1-(2-hydroxypropyl)-3-methyl-hepta-1,2-dienyl] phosphonates with electrophiles that produces 2-oxo-2,5-dihydro-1,2-oxaphospholes due to the participation of the phosphonate neighboring group in the cyclization is described. On the other hand, hept-(1E)-en-1-yl phosphine oxides were prepared as mixtures with 2,5-dihydro-1,2-oxaphosphol-2-ium chlorides in a ratio of about 1:2 by chemo-, regio-, and stereoselective electrophilic addition to the C²-C³-double bond in the allene moiety and subsequent concurrent attack of the external (chloride anion) and internal (phosphine oxide group) nucleophiles. Phosphorylated β -hydroxyallenes were smoothly converted into the corresponding 4-phosphoryl-3,6-dihydro-2H-pyrans by using 5 mol % of coinage metal salt as catalyst in 6-*endo-trig* cycloisomerization reaction.

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INTRODUCTION

Functionalized allenes have attracted a growing attention because of their versatility as key building blocks for organic synthesis. The synthetic potential of functionalized allenes has been explored extensively in recent years, and this has led to the development of novel methods for the construction of a variety of functionalized heterocyclic and carbocyclic systems (Bates and Satcharoen, 2002; Ma, 2007; Hassan, 2007; Pinho e Melo, 2009; Back *et al.*, 2010).

A plethora of methods exists for the construction of hydroxyallenes, including prototropic rearrangement of propargylic alcohols (Enomoto, 1986; Oroshnik, 1953; Phadtare and Zemlicka, 1989), metal-catalyzed nucleophilic addition of propargylic derivatives to aldehydes (Crabbé, 1979; Hoffman and Weldmann, 1985; Boldrini *et al.*, 1987; Corey *et al.*, 1989, 1990; Corey and Jones, 1991; Ma *et al.*, 2002; Ye *et al.*, 2012), Cu(I)-catalyzed reaction of propargylic chlorides with Grignard reagents (Deng *et al.*, 2007; Li, 2009a,b), metal-

catalyzed reaction of propargylic oxiranes with organometallic compounds (Alexakis *et al.*, 1989, 1991; Marshall and Pinney, 1993; Krause *et al.*, 2002, 2004; Deutsch *et al.*, 2007; Aksin-Artok *et al.*, 2011; Poonoth and Krause, 2011) and ketones (Aurrecochea and Solay, 1995; Aurrecochea *et al.*, 1998), and by other methods (Marshall and Tang, 1993; Krause *et al.*, 2012).

There are methods (Mark, 1970; Landor, 1982; Saalfrank and Lurz, 1993; Hashmi, 2004) for the synthesis of phosphorus-containing allenes (phosphonates (Macomber, 1977; Denmark and Marlin, 1991; Cai and Blackburn, 1997; Saalfrank *et al.*, 1999; Kumar *et al.*, 2008, 2009), and phosphine oxides (Nicolaou *et al.*, 1990; Curfin and Okamura, 1990; Grissom and Huang, 1995; Darcel *et al.*, 1996; De Frutos and Echavarren, 1997; Schmittel *et al.*, 2001; Guo *et al.*, 2008; Srinivas *et al.*, 2011) including reactions of α -alkynols with chloride-containing derivatives of phosphorus acids followed by [2,3]-sigmatropic rearrangement. Several diethylphosphono-substituted α -allenenic alcohols were prepared

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by (Brel, 1999, 2006; Brel and Abramkin, 2002) directly from alcohols by Horner-Mark rearrangement of unstable propargylic phosphites.

Transition metal-catalyzed cyclization of functionalized allenes bearing a nucleophilic center has attracted considerable attention in recent years (Zimmer *et al*, 2010). Particularly, the cyclization reactions of allenols catalyzed by Ag(I) (Olsson and Claesson, 1979; Nikam *et al*, 1986; Marshall and Sehon, 1995; Marshall *et al*, 1995), Hg(II) (Gelin *et al*, 1972; Chilot *et al*, 1981), Pd(0) (Kang *et al*, 1998, 1999; Uemura *et al*, 1999), Pd(II) (Ma and Gao 2000, 2002), or Ru(III) (Trost and Pinkerton, 1999, Yoneda *et al*, 2000) have become quite useful methodologies for the synthesis of five-, or six-membered oxygen-containing heterocycles. Krause's group has reported a highly efficient and stereoselective synthesis of 2,5-dihydrofurans by Au(I)- and Au(III)-catalyzed (Krause *et al*, 2005, 2006, 2008) cycloisomerization of α -hydroxyallenes (Krause *et al*, 2001, 2002, 2007). Moreover, the method is not restricted to the cycloisomerization of α -hydroxyallenes to 2,5-dihydrofurans (Krause *et al*, 2001, 2002), rather, it was recently extended by Krause's group to the corresponding endo-cyclization of β -hydroxyallenes (Gockel and Krause, 2006) to the corresponding six-membered *O*-heterocycles. The method of choice, however, is the use of transition metal catalysts since this combines high reactivities and excellent yields with a tolerance to many functional groups.

Acyclic analogs of nucleotides containing an allenic skeleton were prepared by Brel and coworkers (Brel *et al*, 2005a) directly from alcohols by Horner-Mark [2,3]-sigmatropic rearrangement of unstable propargylic phosphites. Intramolecular cyclization of the alkoxides tethered to the allenyl moiety in the presence of AgNO₃ yielded 3,6-dihydro-2H-pyran-4-yl- and 4,5-dihydro-3-furanyl-phosphonates (Brel *et al*, 2005b, 2006).

On the other hand, the literature data on the reactions of phosphorylated allenes with electrophilic reagents reveal that the reactions proceed with cyclization of the allenic system bearing the phosphoryl group (O=P-C=C=C) to give heterocyclic compounds in most cases and the outcome depends on the structure of the starting allenic compound as well as the type of electrophile used (Angelov, 1983; Khusainova and Pudovik 1987; Allabugin and Brel, 1997; Ma, 2009).

As a part of our research program on the chemistry of the bifunctionalized allenes, we required a convenient method to introduce a phosphorus-containing group such as phosphonate or phosphine oxide group as well as a β -hydroxyalkyl group in the first position to the allenic system of double bonds. The above mentioned groups attract increasing attention as useful functionalities in organic synthesis. Of particular interest are the applications of these groups as temporary transformers of chemical reactivity of the allenic system in the synthesis of eventually heterocyclic compounds. In a continuation to our previous reports on the synthesis (Christov *et al*, 2014a) and cyclization reactions (Christov *et al*, 2014b, 2015) of

phosphorylated α -hydroxyallenes, we have found a convenient and efficient method for regioselective synthesis of phosphorylated β -hydroxyallenes by an atom economical [2,3]-sigmatropic rearrangement of the mediated 5-(tetrahydro-2H-pyran-2-yloxy)-hex-2-ynyl phosphite or phosphinite, which to be used as starting materials in the electrophilic cyclization and coinage metal-catalyzed cycloisomerization reactions to 2,5-dihydro-1,2-oxaphospholes and 4-phosphoryl-3,6-dihydro-2H-pyrans respectively.

MATERIALS AND METHODS

General Information

All new synthesized compounds were purified by column chromatography and characterized on the basis of NMR, IR, and microanalytical data. NMR spectra were recorded on DRX Bruker Avance-250 (Bruker BioSpin, Karlsruhe, Germany) (¹H at 250.1 MHz, ¹³C at 62.9 MHz, ³¹P at 101.2 MHz) and Bruker Avance II+600 (Bruker BioSpin GmbH, Karlsruhe, Germany) (¹H at 600.1 MHz, ¹³C at 150.9 MHz, ³¹P at 242.9 MHz) spectrometers for solutions in CDCl₃. All ¹H and ¹³C NMR experiments were measured referring to the signal of internal TMS and ³¹P NMR experiments were measured referring to the signal of external 85% H₃PO₄. *J* values are given in hertz. IR spectra were recorded with an FT-IRAffinity-1 Shimadzu spectrophotometer (Shimadzu, Tokyo, Japan). Elemental analyses were carried out by the Microanalytical Service Laboratory of Faculty of Chemistry and Pharmacy, University of Sofia, Bulgaria, using Vario EL3 CHNS(O) (Elementar Analysensysteme, Hanau, Germany). Column chromatography was performed on Kieselgel F₂₅₄60 (70–230 mesh ASTM, 0.063–0.200 mm, Merck). Et₂O and THF were distilled from Na wire/benzophenone, CH₂Cl₂ was distilled over CaH₂, and other organic solvents used in this study were dried over appropriate drying agents by standard methods and distilled prior to use. All other chemicals used in this study were commercially available and were used without additional purification unless otherwise noted. Reactions were carried out in oven dried glassware under an argon atmosphere and exclusion of moisture. All compounds were checked for purity on TLC plates Kieselgel F₂₅₄ 60 (Merck).

Procedure (Robertson, 1960; Miyashita *et al*, 1977) for Synthesis of the 2-(1-Methyl-but-3-yn-1-yloxy)-tetrahydro-2H-pyran 2

A solution of the pent-4-yn-2-ol **1** (60 mmol) and DHP (3,4-dihydro-2H-pyran) (7.57 g, 90 mmol) in dry methylene chloride (50 mL) containing PPTS (pyridinium p-toluenesulfonate) (1.50 g, 6 mmol) is stirred for 4h at room temperature. Then the reaction was quenched with saturated NaHCO₃ end extracted with methylene chloride. The organic layer was dried over anhydrous sodium sulfate. After evaporation of the solvent, the residue was chromatographed on a column (silica gel, Kieselgel Merck 60 F₂₅₄) with a mixture of ethyl acetate and hexane (3:1) as an eluent. The pure product **2** had the following properties:

2-(1-Methyl-but-3-yn-1-yloxy)-tetrahydro-2H-pyran (2).

This compound was obtained as colourless oil, yield: 87%. R_f 0.55. IR (neat, ν_{\max} , cm^{-1}): 1125 (C-O-C), 2106 (C≡C), 3292 (≡C-H). $^1\text{H-NMR}$ (250.1 MHz): δ_{H} 1.09-1.25, 3.65-3.77, 4.74-4.81 (mmm, 9H, OTHP), 1.27 (d, $J = 7.2$ Hz, 3H, Me-CHO), 2.02 (m, 1H, H-C≡), 2.38, 2.59 (overlapping multiplets, 2H, CH-C≡), 3.80-3.89 (m, 1H, Me-CHO). $^{13}\text{C-NMR}$ (62.9 MHz) δ_{C} 20.1 (CH₂), 22.7 (CH₃), 25.8 (CH₂), 26.4 (CH₂), 31.9 (CH₂), 63.8 (CH₂), 67.0 (CH), 75.2 (CH), 81.2 (C), 95.9 (CH). Anal. Calcd for C₁₀H₁₆O₂ requires: C 71.39, H 9.59. Found: C 71.32, H 9.65.

Procedure for Synthesis of the 5-Methyl-9-(tetrahydro-2H-pyran-2-yloxy)-dec-6-yn-5-ol 5

Ethylmagnesium bromide [prepared from magnesium (1.22 g, 50 mmol) and ethyl bromide (5.50 g, 50 mmol) in dry THF (50 mL)] is added dropwise under stirring to substituted alkynyloxy-tetrahydro-2H-pyran **2** (50 mmol) and then the mixture is refluxed for 2h. The solution of the prepared pentynyl magnesium bromide **3** is added dropwise under stirring to the hexan-2-one **4** (100 mmol). The mixture is refluxed for 24 h and after cooling is hydrolyzed with a saturated aqueous solution of ammonium chloride. The organic layer is separated, washed with water, and dried over anhydrous sodium sulfate. Solvent and the excess of ketone are removed by distillation. Purification of the residue is achieved by column chromatography (silica gel, Kieselgel Merck 60 F₂₅₄) with ethyl acetate-hexane (5:1). The pure product **5** had the following properties:

5-Methyl-9-(tetrahydro-2H-pyran-2-yloxy)-dec-6-yn-5-ol (5).

This compound was obtained as colourless oil, yield: 55%. R_f 0.51. IR (neat, ν_{\max} , cm^{-1}): 1122 (C-O-C), 3408 (OH). $^1\text{H-NMR}$ (250.1 MHz): δ_{H} 0.88 (t, $J = 6.5$ Hz, 3H, Me-(CH₂)₃), 1.15-1.37, 3.60-3.81, 4.73-4.81 (overlapping multiplets, 9H, OTHP), 1.28 (d, $J = 7.1$ Hz, 3H, Me-CHO), 1.30-1.58 (m, 2H, Me-(CH₂)₂), 1.37 (s, 3H, Me-C), 2.54 (s, 1H, OH), 2.57-2.64 (m, 2H, O-CH-CH₂-C≡), 3.83-3.88 (m, 1H, Me-CHO). $^{13}\text{C-NMR}$ (62.9 MHz) δ_{C} 14.7 (CH₃), 20.1 (CH₂), 22.7 (CH₃), 24.3 (CH₂), 24.7 (CH₂), 25.1 (CH₂), 25.6 (CH₂), 30.0 (CH₃), 31.8 (CH₂), 45.3 (CH₂), 63.9 (CH₂), 68.1 (C), 75.7 (CH), 78.4 (C), 82.3 (C), 96.1 (CH). Anal. Calcd for C₁₆H₂₈O₃ requires: C 71.60, H 10.52. Found: C 71.66, H 10.44.

Procedure for Synthesis of the Dimethyl 3-Methyl-1-[2-(tetrahydro-2H-pyran-2-yloxy)-propyl]-hepta-1,2-dienephosphonate 7

To a solution of phosphorus trichloride (2.75 g, 20 mmol) and triethylamine (2.23 g, 22 mmol) in dry diethyl ether (60 mL) at -70°C was added dropwise with stirring a solution of the 5-methyl-9-(tetrahydro-2H-pyran-2-yloxy)-dec-6-yn-5-ol **5** (20 mmol) in the same solvent (20 mL). After 30 min stirring at the same condition a solution of pyridine (3.16 g, 44 mmol) and of methanol (1.28 g, 40 mmol) in dry diethyl ether (50 mL) were added. The reaction mixture was stirred for an hour at the same temperature and for 10 hours at room temperature. The mixture was then washed with water, 2N HCl, extracted

with ether, washed with saturated NaCl, and dried over anhydrous sodium sulfate. After evaporation of the solvent, the residue was chromatographed on a column (silica gel, Kieselgel Merck 60 F₂₅₄) with a mixture of ethyl acetate and hexane (10:1) as an eluent to give the pure product **7** as an oil, which had the following properties:

Dimethyl 3-Methyl-1-[2-(tetrahydro-2H-pyran-2-yloxy)-propyl]-hepta-1,2-dienephosphonate (7).

This compound was obtained as orange oil, yield: 71%. R_f 0.45. IR (neat, ν_{\max} , cm^{-1}): 1118 (C-O-C), 1258 (P=O), 1955 (C=C=C). $^1\text{H-NMR}$ (600.1 MHz): δ_{H} 0.90 (t, $J = 7.3$ Hz, 3H, Me-(CH₂)₃), 1.17 (d, $J = 6.0$ Hz, 3H, Me-CHO), 1.20-1.23, 1.24-1.29, 1.42-1.47, 3.84-3.95, 4.90-4.94 (overlapping multiplets, 9H, OTHP), 1.32-1.38 (m, 2H, Me-CH₂(CH₂)₂), 1.51-1.57 (m, 2H, Me-CH₂CH₂CH₂), 1.76 (d, $J = 7.0$ Hz, 3H, Me-C=), 2.14-2.42 (m, 2H, O-CH-CH₂-C=), 2.35-2.42 (m, 2H, Me-(CH₂)₂CH₂), 3.73 (d, $J = 11.1$ Hz, 3H, MeO), 4.67-4.71 (m, 1H, Me-CHO). $^{13}\text{C-NMR}$ (150.9 MHz) δ_{C} 13.9 (CH₃), 18.8 ($J = 5.3$ Hz, CH₃), 19.7 (CH₂), 20.0 ($J = 7.8$ Hz, CH₃), 20.4 (CH₂), 25.3 (CH₂), 29.4 (CH₂), 32.0 (CH₂), 36.1 ($J = 6.7$ Hz, CH₂), 36.9 ($J = 8.9$ Hz, CH₂), 52.8 ($J = 6.4$ Hz, CH₃), 62.9 ($J = 12.1$ Hz, CH), 63.7 (CH₂), 88.4 ($J = 191.8$ Hz, C), 96.2 (CH), 102.4 ($J = 13.2$ Hz, C), 208.3 ($J = 5.1$ Hz, C). $^{31}\text{P-NMR}$ (242.9 MHz): δ_{P} 21.9. Anal. Calcd for C₁₈H₃₃O₅P requires: C 59.98, H 9.23. Found: C 59.93, H 9.16.

Procedure for the Synthesis of the 2-(3-Diphenylphosphinoyl-1,5-dimethyl-nona-3,4-dienyloxy)-tetrahydro-2H-pyran 9

To a solution of the 5-methyl-9-(tetrahydro-2H-pyran-2-yloxy)-dec-6-yn-5-ol **5** (20 mmol) and triethylamine (2.23 g, 22 mmol) in dry diethyl ether (60 mL) at -70°C, a solution of freshly distilled diphenylchloro phosphine (4.41 g, 20 mmol) in the same solvent (20 mL) was added dropwise with stirring. The reaction mixture was stirred for an hour at the same temperature and for 8 h at room temperature and then washed with water, 2N HCl, extracted with diethyl ether, and the extract was washed with saturated NaCl, and dried over anhydrous sodium sulfate. The solvent was removed using a rotatory evaporator, and the residue was purified by column chromatography on a silica gel (Kieselgel Merck 60 F₂₅₄) with ethyl acetate-hexane (10:1) to give the pure product **9** as an oil, which had the following properties:

2-(3-Diphenylphosphinoyl-1,5-dimethyl-nona-3,4-dienyloxy)-tetrahydro-2H-pyran (9).

This compound was obtained as yellow oil, yield: 79%. R_f 0.43. IR (neat, ν_{\max} , cm^{-1}): 1122 (C-O-C), 1156 (P=O), 1436, 1490 (Ph), 1951 (C=C=C). $^1\text{H-NMR}$ (600.1 MHz): δ_{H} 0.81 (t, $J = 7.3$ Hz, 3H, Me-(CH₂)₃), 1.11-1.16, 1.23-1.28, 1.44-1.49, 3.86-3.94, 4.88-4.93 (overlapping multiplets, 9H, OTHP), 1.17 (d, $J = 6.1$ Hz, 3H, Me-CHO), 1.35-1.39 (m, 2H, Me-CH₂(CH₂)₂), 1.49-1.55 (m, 2H, Me-CH₂CH₂CH₂), 1.79 (d, $J = 7.1$ Hz, 3H, Me-C=), 2.04-2.16 (m, 2H, O-CH-CH₂-C=), 2.29-2.36 (m, 2H, Me-(CH₂)₂CH₂), 4.66-4.70 (m, 1H, Me-CHO), 7.37-7.88 (m, 10H, 2Ph). $^{13}\text{C-NMR}$ (150.9 MHz) δ_{C} 13.9 (CH₃), 19.1 ($J = 5.4$ Hz, CH₃), 19.8 ($J = 8.0$ Hz, CH₃), 21.1

(CH₂), 21.3 (CH₂), 25.4 (CH₂), 29.3 (CH₂), 32.1 (CH₂), 35.1 (*J* = 6.2 Hz, CH₂), 35.6 (*J* = 8.7 Hz, CH₂), 62.7 (*J* = 12.2 Hz, CH), 62.9 (CH₂), 93.8 (*J* = 103.6 Hz, C), 96.2 (CH), 102.9 (*J* = 13.2 Hz, C), 127.7-133.2 (2Ph), 207.9 (*J* = 6.6 Hz, C). ³¹P-NMR (242.9 MHz): δ_P 32.6. Anal. Calcd for C₂₈H₃₇O₃P requires: C 74.31, H 8.24. Found: C 74.39, H 8.20.

Procedure for the Synthesis of the Dimethyl [1-(2-Hydroxypropyl)-3-methyl-hepta-1,2-dienyl]phosphonate 10 and the 4-Diphenylphosphinoyl-6-methyl-deca-4,5-dien-2-ol 11

A solution of the dimethyl 3-methyl-1-[2-(tetrahydro-2H-pyran-2-yloxy)-propyl]-hepta-1,2-dienephosphonate **7** or the 2-(3-diphenylphosphinoyl-1,5-dimethyl-nona-3,4-dienyloxy) tetrahydro-2 H-pyran **9** (5 mmol) and PPTS (0.5 mmol) in ethanol (10 mL) was stirred at room temperature for 6 h.

The mixture was then washed with water, extracted with methylene chloride and dried over anhydrous sodium sulfate. After evaporation of the solvent, the residue was chromatographed on a column (silica gel, Kieselgel Merck 60 F₂₅₄) with a mixture of ethyl acetate and hexane (10:1) as an eluent to give the pure products **10** or **11** as oils, which had the following properties:

Dimethyl [1-(2-Hydroxypropyl)-3-methyl-hepta-1,2-dienyl]phosphonate (10)

This compound was obtained as yellow oil, yield: 83%. R_f 0.59. IR (neat, ν_{max}, cm⁻¹): 1259 (P=O), 1951 (C=C=C), 3420 (OH). ¹H-NMR (600.1 MHz): δ_H 0.91 (t, *J* = 7.3 Hz, 3H, Me-(CH₂)₃), 1.22 (d, *J* = 6.4 Hz, 3H, Me-CHO), 1.33-1.38 (m, 2H, Me-CH₂(CH₂)₂), 1.41-1.47 (m, 2H, Me-CH₂CH₂CH₂), 1.78 (d, *J* = 6.9 Hz, 3H, Me-C=), 2.03-2.06 (m, 2H, O-CH-CH₂-C=), 2.27-2.33 (m, 2H, Me-(CH₂)₂CH₂), 2.88 (s, 1H, OH), 3.74 (d, *J* = 11.1 Hz, 3H, MeO), 4.59-4.62 (m, 1H, Me-CHO). ¹³C-NMR (150.9 MHz) δ_C 13.8 (CH₃), 18.0 (*J* = 5.8 Hz, CH₃), 22.2 (CH₂), 22.7 (*J* = 7.9 Hz, CH₃), 29.3 (CH₂), 33.0 (*J* = 6.6 Hz, CH₂), 39.4 (*J* = 12.1 Hz, CH₂), 52.9 (*J* = 6.3 Hz, CH₃), 66.9 (*J* = 9.7 Hz, CH), 88.6 (*J* = 190.3 Hz, C), 102.3 (*J* = 15.9 Hz, C), 208.1 (*J* = 5.2 Hz, C). ³¹P-NMR (242.9 MHz): δ_P 22.6. Anal. Calcd for C₁₃H₂₅O₄P requires: C 56.51, H 9.12. Found: C 56.43, H 9.19.

4-Diphenylphosphinoyl-6-methyl-deca-4,5-dien-2-ol (11)

This compound was obtained as orange oil, yield: 88%. R_f 0.58. IR (neat, ν_{max}, cm⁻¹): 1167 (P=O), 1436, 1491 (Ph), 1949 (C=C=C), 3401 (OH). ¹H-NMR (600.1 MHz): δ_H 0.82 (t, *J* = 7.2 Hz, 3H, Me-(CH₂)₃), 1.20 (d, *J* = 6.4 Hz, 3H, Me-CHO), 1.40-1.44 (m, 2H, Me-CH₂(CH₂)₂), 1.46-1.51 (m, 2H, Me-CH₂CH₂CH₂), 1.80 (d, *J* = 7.1 Hz, 3H, Me-C=), 2.04-2.10 (m, 2H, O-CH-CH₂-C=), 2.30-2.36 (m, 2H, Me-(CH₂)₂CH₂), 2.90 (s, 1H, OH), 4.58-4.61 (m, 1H, Me-CHO), 7.35-7.89 (m, 10H, 2Ph). ¹³C-NMR (150.9 MHz) δ_C 13.9 (CH₃), 18.7 (*J* = 5.9 Hz, CH₃), 21.1 (CH₂), 22.9 (*J* = 8.0 Hz, CH₃), 29.2 (CH₂), 32.9 (*J* = 6.4 Hz, CH₂), 39.5 (*J* = 12.4 Hz, CH₂), 67.3 (*J* = 9.8 Hz, CH), 94.6 (*J* = 102.4 Hz, C), 102.2 (*J* = 15.7 Hz, C), 128.0-

132.5 (2Ph), 209.0 (*J* = 5.4 Hz, C). ³¹P-NMR (242.9 MHz): δ_P 34.5. Anal. Calcd for [C₂₃H₂₉O₂P] requires: C 74.98, H 7.93. Found: C 75.04, H 7.88.

General Procedure for the Reactions of the Phosphorylated S-Hydroxyallenes 7, 9-11 with Electrophilic Reagents

To a solution of the phosphorylated β-hydroxyallene with protected (**7** or **9**) or unprotected (**10** or **11**) hydroxy group (3.0 mmol) in dry dichloromethane (10 mL) at -20 °C was added dropwise with stirring a solution of electrophilic reagent (sulfuryl chloride, bromine, phenylsulfenyl chloride or benzeneselenenyl chloride) (3.6 mmol) in the same solvent (10 mL). The reaction mixture was stirred at the same temperature for 2 hours and 2 hours at room temperature.

After evaporation of the solvent, the residue was chromatographed on a silica gel column (ethyl acetate and hexane 4:1 or 5:1) as an eluent to give the pure products **12-14** as oils, which had the following properties:

2[2-(4-Bromo-5-butyl-2-methoxy-5-methyl-2-oxo-2,5-dihydro-1,2-oxaphosphol-3-yl)-1-methylethyl]-tetrahydro-2H-pyran (12a)

This compound was obtained as yellow oil, yield: 80%. R_f 0.51. IR (neat, ν_{max}, cm⁻¹): 1016 (C-O-P), 1121 (C-O-C), 1585 (P=O), 1584 (C=C). ¹H-NMR (600.1 MHz): δ_H 0.92 (t, *J* = 6.2 Hz, 3H, Me-(CH₂)₃), 1.12-1.26, 3.32-3.71, 4.61-4.69 (overlapping multiplets, 9H, OTHP), 1.29 (d, *J* = 6.3 Hz, 3H, Me-CH), 1.24-1.35, 1.42-1.49, 1.72-1.88 (overlapping multiplets, 6H, (CH₂)₃-Me), 1.57 (s, 3H, Me-C), 2.38-2.55 (m, 2H, CH₂), 3.77 (d, *J* = 11.6 Hz, 3H, MeO), 4.50-4.60 (m, 1H, Me-CH). ¹³C-NMR (150.9 MHz) δ_C 14.0, 19.6, 22.7 (*J* = 7.5 Hz), 23.2 (*J* = 4.7 Hz), 23.3, 25.3, 25.7 (*J* = 7.8 Hz), 31.1, 35.6 (*J* = 5.8 Hz), 39.4 (*J* = 7.9 Hz), 52.5 (*J* = 15.2 Hz), 62.7, 70.0 (*J* = 8.1 Hz), 89.5 (*J* = 9.8 Hz), 96.3, 125.7 (*J* = 156.0 Hz), 142.3 (*J* = 51.1 Hz). ³¹P-NMR (242.9 MHz): δ_P 33.5. Anal. Calcd for C₁₇H₃₀O₅PBr requires: C 48.01, H 7.11. Found: C 48.05, H 7.13.

1-(5-Butyl-2-methoxy-5-methyl-2-oxo-4-phenylsulfenyl-2,5-dihydro-1,2-oxaphosphol-3-yl)propan-2-ol (12b)

This compound was obtained as yellow oil, yield: 87%. R_f 0.55. IR (neat, ν_{max}, cm⁻¹): 1024 (C-O-P), 1259 (P=O), 1581 (C=C), 3410 (OH). ¹H-NMR (600.1 MHz): δ_H 0.91 (t, *J* = 6.3 Hz, 3H, Me-(CH₂)₃), 1.26 (d, *J* = 6.1 Hz, 3H, Me-CH), 1.27-1.34, 1.60-1.68, 1.86-2.04 (overlapping multiplets, 6H, (CH₂)₃-Me), 1.46 (s, 3H, Me-C), 2.55-2.67 (m, 2H, CH₂), 3.28 (s, 1H, OH), 3.73 (d, *J* = 11.4 Hz, 3H, MeO), 3.92-3.98 (m, 1H, Me-CH), 7.14-7.41 (m, 5H, Ph). ¹³C-NMR (150.9 MHz) δ_C 14.1, 23.0, 23.6 (*J* = 4.6 Hz), 28.2 (*J* = 4.7 Hz), 38.5 (*J* = 5.8 Hz), 39.4 (*J* = 7.8 Hz), 52.4 (*J* = 15.1 Hz), 65.8 (*J* = 8.1 Hz), 88.2 (*J* = 9.9 Hz), 126.7-134.5, 126.8 (*J* = 100.9 Hz), 164.2 (*J* = 15.4 Hz). ³¹P-NMR (242.9 MHz): δ_P 34.5. Anal. Calcd for C₁₈H₂₇O₄PS requires: C 58.36, H 7.35. Found: C 58.30, H 7.37.

5-Butyl-2,2-diphenyl-5-methyl-4-phenylselenenyl-3-[(tetrahydro-2H-pyran-2-yloxy)prop-2-yl]-2,5-dihydro-1,2-oxaphosphol-2-onium chloride (13a)

This compound was obtained as yellow oil, yield: 51%. R_f 0.59. IR (neat, ν_{max} , cm^{-1}): 1112 (C-O-C), 1439, 1477 (Ph), 1577 (C=C). 1H -NMR (600.1 MHz): δ_H 0.91 (t, $J = 6.3$ Hz, 3H, $\underline{Me}-(CH_2)_3$), 1.10-1.26, 3.47-3.66, 4.55-4.60 (overlapping multiplets, 9H, OTHP), 1.25-1.30, 1.47-1.53, 1.80-2.00 (overlapping multiplets, 6H, $(CH_2)_3$ -Me), 1.36 (d, $J = 6.2$ Hz, 3H, $\underline{Me}-CH$), 1.60 (s, 3H, Me-C), 2.84-3.01 (m, 2H, CH_2), 3.77-3.89 (m, 1H, $\underline{Me}-CH$), 7.15-7.73 (m, 15H, 3Ph). ^{13}C -NMR (150.9 MHz) δ_C 14.0, 19.8, 22.3 ($J = 10.1$ Hz), 22.6 ($J = 4.7$ Hz), 25.1 ($J = 7.9$ Hz), 25.6, 31.2, 35.1 ($J = 6.0$ Hz), 40.2 ($J = 7.8$ Hz), 63.2, 75.8 ($J = 8.0$ Hz), 95.5 ($J = 9.9$ Hz), 96.1, 127.6-137.8, 152.4 ($J = 96.1$ Hz), 168.0 ($J = 85.3$ Hz). ^{31}P -NMR (242.9 MHz): δ_P 82.8. Anal. Calcd for $C_{34}H_{43}O_3PSeCl$ requires: C 63.30, H 6.72. Found: C 63.37, H 6.70.

3-Methyl-3-chloro-2-phenylselenenyl-1-[2-(tetrahydro-2H-pyran-2-yloxy)prop-2-yl]hept-(1E)-en-1-yl diphenyl phosphine oxide (14a)

This compound was obtained as orange oil, yield: 25%. R_f 0.42. IR (neat, ν_{max} , cm^{-1}): 1122 (C-O-C), 1150 (P=O), 1437, 1478 (Ph), 1592 (C=C). 1H -NMR (600.1 MHz): δ_H 0.87 (t, $J = 6.0$ Hz, 3H, $\underline{Me}-(CH_2)_3$), 1.11-1.26, 3.52-3.63, 4.58-4.64 (overlapping multiplets, 9H, OTHP), 1.28 (d, $J = 6.1$ Hz, 3H, $\underline{Me}-CH$), 1.31-1.39, 1.37-1.45, 2.36-2.43 (overlapping multiplets, 6H, $(CH_2)_3$ -Me), 1.77 (s, 3H, Me-C), 2.56-2.65 (m, 2H, CH_2), 3.95-4.05 (m, 1H, $\underline{Me}-CH$), 7.38-7.73 (m, 15H, 3Ph). ^{13}C -NMR (150.9 MHz) δ_C 14.1, 19.9, 22.5, 22.8 ($J = 5.1$ Hz), 25.5, 25.7 ($J = 5.9$ Hz), 29.1 ($J = 5.1$ Hz), 31.3, 39.8 ($J = 5.8$ Hz), 42.6 ($J = 4.7$ Hz), 63.0, 74.1 ($J = 7.8$ Hz), 81.3 ($J = 8.2$ Hz), 95.8, 127.8 ($J = 100.5$ Hz), 128.4-138.9, 152.7 ($J = 85.2$ Hz). ^{31}P -NMR (242.9 MHz): δ_P 21.9. Anal. Calcd for $C_{34}H_{42}ClO_3PSe$ requires: C 63.40, H 6.57. Found: C 63.35, H 6.52.

5-Butyl-4-chloro-2,2-diphenyl-5-methyl-3-(2-hydroxypropyl)-2,5-dihydro-1,2-oxaphosphol-2-onium chloride (13b)

This compound was obtained as yellow oil, yield: 57%. R_f 0.60. IR (neat, ν_{max} , cm^{-1}): 1440, 1495 (Ph), 1585 (C=C), 3397 (OH). 1H -NMR (600.1 MHz): δ_H 0.91 (t, $J = 6.3$ Hz, 3H, $\underline{Me}-(CH_2)_3$), 1.13-1.21, 1.26-1.37, 2.27-2.46 (overlapping multiplets, 6H, $(CH_2)_3$ -Me), 1.29 (d, $J = 6.1$ Hz, 3H, $\underline{Me}-CH$), 1.68 (s, 3H, Me-C), 2.66 (s, 1H, OH), 2.77-2.88 (m, 2H, CH_2), 4.05-4.14 (m, 1H, $\underline{Me}-CH$), 7.79-8.24 (m, 10H, 2Ph). ^{13}C -NMR (150.9 MHz) δ_C 14.1, 22.2 ($J = 4.9$ Hz), 23.0, 23.6 ($J = 4.4$ Hz), 27.3 ($J = 8.1$ Hz), 34.8 ($J = 5.9$ Hz), 38.6 ($J = 8.0$ Hz), 70.4 ($J = 7.8$ Hz), 94.2 ($J = 9.9$ Hz), 124.7 ($J = 97.4$ Hz), 125.4-136.1, 165.7 ($J = 39.6$ Hz). ^{31}P -NMR (242.9 MHz): δ_P 84.0. Anal. Calcd for $C_{23}H_{29}O_2PCl_2$ requires: C 62.88, H 6.65. Found: C 62.86, H 6.66.

5,6-Dichloro-4-diphenylphosphoryl-6-methyl-dec-(4E)-en-2-ol (14b)

This compound was obtained as orange oil, yield: 29%. R_f 0.44. IR (neat, ν_{max} , cm^{-1}): 1171 (P=O), 1438, 1485 (Ph), 1606

(C=C), 3310 (OH). 1H -NMR (600.1 MHz): δ_H 0.89 (t, $J = 6.1$ Hz, 3H, $\underline{Me}-(CH_2)_3$), 1.29 (d, $J = 6.3$ Hz, 3H, $\underline{Me}-CH$), 1.36-1.45, 1.46-1.51, 2.28-2.45 (overlapping multiplets, 6H, $(CH_2)_3$ -Me), 1.83 (s, 3H, Me-C), 2.47-2.57 (m, 2H, CH_2), 3.27 (s, 1H, OH), 4.03-4.13 (m, 1H, $\underline{Me}-CH$), 7.54-7.90 (m, 10H, 2Ph). ^{13}C -NMR (150.9 MHz) δ_C 14.0, 22.6 ($J = 5.1$ Hz), 22.7, 27.4, 29.3 ($J = 4.9$ Hz), 38.7 ($J = 6.0$ Hz), 43.0 ($J = 4.8$ Hz), 73.1 ($J = 7.9$ Hz), 78.0 ($J = 8.0$ Hz), 124.8 ($J = 100.1$ Hz), 129.3-135.8, 152.7 ($J = 40.1$ Hz). ^{31}P -NMR (242.9 MHz): δ_P 21.6. Anal. Calcd for $C_{23}H_{29}Cl_2O_2P$ requires: C 62.88, H 6.65. Found: C 62.84, H 6.60.

General Procedure for Coinage Metal-catalyzed Cycloisomerization of the 1-(2-Hydroxypropyl)-allenephosphonate 10 and 1-(2-Hydroxypropyl) Allenyl Phosphine Oxide 11

Coinage metal salt catalyst (0.15 mmol) was added to a solution of the 1-(2-hydroxypropyl)-allenephosphonate **10** or 1-(2-hydroxypropyl) allenyl phosphine oxide **11** (3.0 mmol) in dry dichloromethane (10 mL). The mixture was stirred at room temperature and in the dark for the hours indicated in the **Table 1**. Saturated sodium chloride solution was added to precipitate the silver ions. The product was extracted by chloroform. The organic layer was dried over anhydrous sodium sulfate. After evaporation of the solvent, the residue was chromatographed on a column (silica gel, Kieselgel Merck 60 F₂₅₄) with a mixture of ethyl acetate and hexane (6:1) as an eluent to give the pure products **15** as oils, which had the following properties:

Dimethyl (6-butyl-2,6-dimethyl-3,6-dihydro-2H-pyran-4-yl)phosphonate (15a)

This compound was obtained as orange oil, yield: 21-66% (Table 1). R_f 0.61. IR (neat, ν_{max} , cm^{-1}): 1126 (C-O-C), 1242 (P=O), 1624 (C=C). 1H -NMR (600.1 MHz): δ_H 0.89 (t, $J = 7.1$ Hz, 3H, $\underline{Me}-(CH_2)_3$), 1.28-1.38, 1.44-1.60 (overlapping multiplets, 6H, $(CH_2)_3$ -Me), 1.33 (s, 3H, Me-C), 1.46 (d, $J = 11.6$ Hz, 3H, $\underline{Me}-CH$), 2.31-2.57 (m, 2H, CH_2), 3.69-3.71 (m, 1H, $\underline{Me}-CH$), 3.77 (d, $J = 11.3$ Hz, 6H, 2MeO), 6.56 (d, $J = 11.6$ Hz, 1H, HC=). ^{13}C -NMR (150.9 MHz) δ_C 14.0, 19.7, 23.0, 26.9, 27.5 ($J = 4.5$ Hz), 28.7 ($J = 15.0$ Hz), 39.4 ($J = 4.0$ Hz), 52.5 ($J = 14.7$ Hz), 66.7 ($J = 7.8$ Hz), 67.9 ($J = 7.8$ Hz), 67.9 ($J = 7.9$ Hz), 127.8 ($J = 188.7$ Hz), 150.6 ($J = 10.1$ Hz). ^{31}P -NMR (242.9 MHz): δ_P 17.1. Anal. Calcd for $C_{13}H_{25}O_4P$ requires: C 56.51, H 9.12. Found: C 56.55, H 9.15.

(6-Butyl-2,6-dimethyl-3,6-dihydro-2H-pyran-4-yl) diphenyl phosphine oxide (15b)

This compound was obtained as orange oil, yield: 19-63% (Table 1). R_f 0.58. IR (neat, ν_{max} , cm^{-1}): 1120 (C-O-C), 1173 (P=O), 1438, 1495 (Ph), 1621 (C=C). 1H -NMR (600.1 MHz): δ_H 0.83 (t, $J = 6.2$ Hz, 3H, $\underline{Me}-(CH_2)_3$), 1.24-1.30, 1.31-1.57 (overlapping multiplets, 6H, $(CH_2)_3$ -Me), 1.28 (d, $J = 11.6$ Hz, 3H, $\underline{Me}-CH$), 1.29 (s, 3H, Me-C), 2.31-2.59 (m, 2H, CH_2), 3.50-3.60 (m, 1H, $\underline{Me}-CH$), 6.65 (d, $J = 12.4$ Hz, 1H, HC=), 7.47-7.78 (m, 10H, 2Ph). ^{13}C -NMR (150.9 MHz) δ_C 14.0, 21.5, 23.6, 25.8, 26.8 ($J = 4.6$ Hz), 38.1 ($J = 14.8$ Hz), 39.3 ($J = 4.7$ Hz), 68.8 ($J = 7.8$ Hz), 70.8 ($J = 8.0$ Hz), 130.0 ($J = 179.9$

Hz), 128.5-134.6, 149.2 ($J = 8.8$ Hz). ^{31}P -NMR (242.9 MHz): δ_{P} 21.6. Anal. Calcd for $\text{C}_{23}\text{H}_{29}\text{O}_2\text{P}$ requires: C 74.98, H 7.93. Found: C 74.95, H 7.95.

RESULTS AND DISCUSSION

Synthesis of Phosphorylated S-Hydroxyallenes

Our strategy for the synthesis of the phosphorylated β -hydroxyallenes, using our experience on the preparation of the phosphorylated α -hydroxyallenes (Christov *et al*, 2014a), relies on the well-precedented [2,3]-sigmatropic shift of propargylic phosphites to allenephosphonates (Macomber, 1977; Denmark and Marlin, 1991; Cai and Blackburn, 1997; Saalfrank *et al*, 1999; Kumar *et al*, 2008, 2009) and propargylic phosphinites to allenyl phosphine oxides (Nicolaou *et al*, 1990; Curfin and Okamura, 1990; Grissom and Huang, 1995; Darcel *et al*, 1996; De Frutos and Echavarren, 1997; Schmittel *et al*, 2001; Guo *et al*, 2008; Srinivas *et al*, 2011).

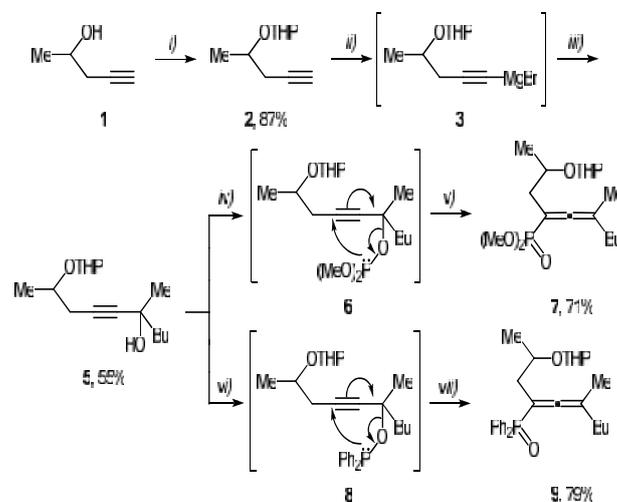
In order to assess this approach towards the target 1,1-bifunctionalized allenes, a range of the phosphorylated β -hydroxyallenes **7**, **9**, **10**, and **11**, was prepared by the following four-step procedure (Christov *et al*, 2014b) including i) protection of hydroxy group in the pent-4-yn-2-ol **1**; ii) subsequent reaction with Grignard reagent and butylmethyl ketone to give the 6-methyl-dec-4-yne-2,6-diol **5** with protected hydroxy group at position 2; iii) interaction with chloride of phosphorus acid in the presence of a base; and finally iv) [2,3]-sigmatropic rearrangement of the mediated protected propargyl phosphite or phosphinite.

As a starting point for our investigation, we first examined the protection of hydroxy group in the pent-4-yn-2-ol **1** with DHP in the presence of PPTS (Robertson, 1960; Miyashita *et al*, 1977) (Scheme 1). Thus, the formed 2-(1-methyl-but-3-ynyloxy)-tetrahydro-2H-pyran **2** was isolated by column chromatography with excellent yield (87%). Reaction of the protected pent-4-yn-2-ol **2** with ethyl magnesium bromide and subsequent dropwise addition of the *in situ* generated pentynyl magnesium bromide **3** to the hexan-2-one **4** and reflux for 24 hours gives the 5-methyl-9-(tetrahydro-2H-pyran-2-yloxy)-dec-6-yn-5-ol **5**, which is stable and was isolated by column chromatography in 55% yield.

With the required dec-4-yne-2,6-diol **5** with protected hydroxy group at 2 position in hand, we were then able to investigate the proposed reactions with the corresponding chloro-containing phosphorus reagents such as dimethyl chlorophosphite and chlorodiphenyl phosphine in the presence of a base and subsequent [2,3]-sigmatropic rearrangement of the mediated 4-(tetrahydro-2H-pyran-2-yloxy)-propargyl phosphite **6** or phosphinite **8**.

In the first instance, the dimethyl 1-(tetrahydro-2H-pyran-2-yloxy)-hepta-1,2-dienephosphonates **7** can be readily prepared *via* an atom economical 2,3-sigmatropic rearrangement of the 4-(tetrahydro-2H-pyran-2-yloxy)-propargyl phosphite **6**, intermediate formed by reaction of the 9-(tetrahydro-2H-

pyran-2-yloxy)-dec-6-yn-5-ol **5** with dimethyl chlorophosphite, prepared *in situ* from phosphorus trichloride in the presence of triethylamine and 2 *equiv* of methanol and 2 *equiv* of pyridine, according to Scheme 1.



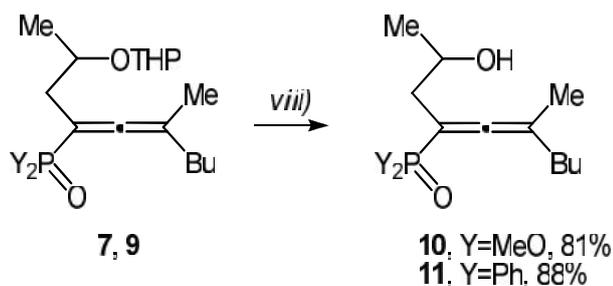
Scheme 1 Synthesis of the phosphorylated β -hydroxyallenes with protected hydroxy group **7** and **9** by 2,3-sigmatropic rearrangement of the propargyl phosphite **6** and phosphinite **8**.

Reagents and Conditions: i) DHP (1.5 eq), PPTS (0.1 eq), CH_2Cl_2 , rt, 4h, distillation; ii) EtMgBr (1 eq), THF, reflux, 2h; iii) dropwise addition of **3** to MeC(O)Bu **4** (2 eq), reflux, 24h, column chromatography; iv) PCl_3 (1 eq), Et_3N (1.1 eq), Et_2O , -70°C , 30 min stirring, pyridine (2.2 eq), MeOH (2 eq), Et_2O , -70°C ; v) [2,3- σ]-rearrangement, -70°C , 1h, rt, 10h; vi) Ph_2PCl (1 eq), Et_3N (1.1 eq), Et_2O , -70°C , 1h, rt, 10h, column chromatography; vii) [2,3- σ]-rearrangement, -70°C , 1h, rt, 8h, column chromatography.

Next, the reaction of the 9-(tetrahydro-2H-pyran-2-yloxy)-dec-6-yn-5-ol **5** with chlorodiphenyl phosphine in the presence of triethylamine at -70°C gave the expected 2-(3-diphenylphosphinoyl-nona-3,4-dienyloxy)-tetrahydro-2H-pyran **9** in very good yield (79%) as a result of [2,3]-sigmatropic rearrangement of the 4-(tetrahydro-2H-pyran-2-yloxy)-propargyl phosphinite **8** for 8 hours at room temperature, according to the reaction sequence outlined in Scheme 1.

A new family of phosphorylated β -hydroxyallenes with protected hydroxy group **7** and **9** were synthesized *via* an atom economical and regioselective [2,3]-sigmatropic rearrangement of the intermediate formed propargyl phosphite **6** or phosphinite **8** in the reaction of the protected dec-6-yn-5-ol **5** with dimethylchloro phosphite or chlorodiphenyl phosphine in the presence of triethylamine. Compounds **7** and **9** were stable enough to be handled at ambient temperature.

The hydroxy group was deprotected by stirring the ethanol solution of the protected hydroxypropyl-allenephosphonate **7** and hydroxypropyl-allenyl phosphine oxide **9** in the presence of 0.1 *equiv* PPTS at room temperature for 6 hours, according to Scheme 2.



Scheme 2 Synthesis of the phosphorylated β -hydroxyallenes **10** and **11**

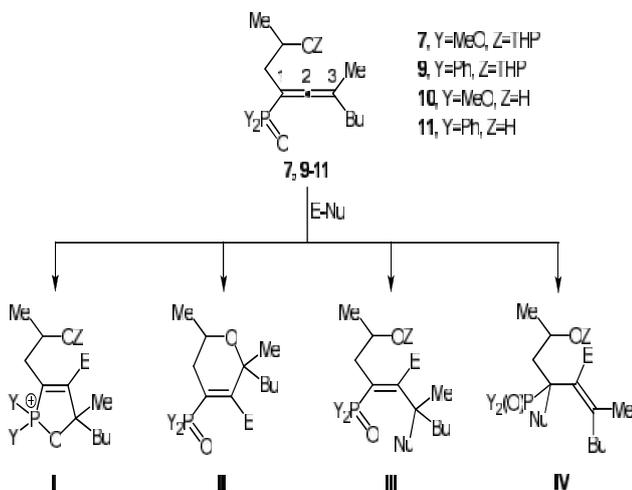
Reagents and Conditions: *viii*) PPTS (0.1 eq), EtOH, rt, 6h, stirring, column chromatography.

After a conventional work-up, all allenic products **7**, **9**, **10**, and **11** were isolated as stable yellow or orange oils by column chromatography and identified by ^1H , ^{13}C , and ^{31}P NMR and IR spectra as well as by elemental analysis.

Phosphorylated β -hydroxyallenes **7**, **9-11** isolated in preparative amounts allowed us to study their chemical behavior in the reactions with electrophilic reagents and the coinage metal-catalyzed cycloisomerization. The present paper is a part of our long-term objective to investigate both the advantages and the limitations of the electrophilic cyclization and cycloisomerization reactions of the phosphorylated β -hydroxyallenes **7**, **9-11**.

Electrophilic Cyclization of Phosphorylated β -Hydroxyallenes

It is necessary to draw attention to the fact that conceptually two distinct modes of cyclization of the phosphorylated β -hydroxyallenes are possible. They depend on the electrophilic atom that forms a new bond with the central carbon of the allenic system, which seems likely (Angelov, 1983; Khusainova and Pudovik, 1987; Allabugin and Brel, 1997; Ma, 2009; Brel, 2006). It is evident that these pathways are closely connected with the intramolecular neighboring group participation of the phosphoryl and/or the hydroxyalkyl groups as internal nucleophile(s) in the final step of the cyclization.

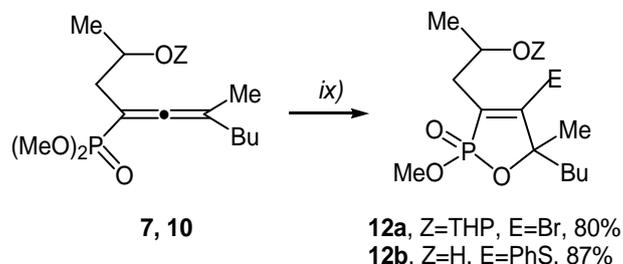


Scheme 3 Probable products of the electrophilic reaction of the phosphorylated β -hydroxyallenes **7**, **9-11**.

Besides the 5-*endo-trig* cyclization (Baldwin, 1976) to the 2,5-dihydro-1,2-oxaphospholes **I** or 6-*endo-trig* cyclization (Baldwin, 1976) to the 4-phosphoryl-3,6-dihydro-2H-pyrans **II**, the electrophilic addition might afford the 2,3-adducts **III** and/or the 3,2-adducts **IV** (Scheme 3).

We started the present study with the reaction of the dimethyl 1-(2-hydroxy-propyl)-3-methyl-hepta-1,2-dienephosphonates with protected (**7**) or unprotected (**10**) hydroxy group with bromine or benzenesulfonyl chloride (Scheme 4). We conducted the reactions under the optimized reaction conditions determined in the similar reactions of the phosphorylated α -hydroxyallenes earlier (Christov *et al*, 2014b) – solvent CH_2Cl_2 at -20°C using 1.0 *equiv* of the allenephosphonate and 1.2 *equiv* of the electrophilic reagent.

We have to say that the reaction in the favour of 5-*endo-trig* mode afforded the 2-methoxy-2-oxo-2,5-dihydro-1,2-oxaphospholes **12a,b** to have very good yields (80 and 87 %) and it does not depend on the nature of the substituent on the hydroxy group, as a result of the neighboring group participation of phosphonate group in the cyclization.



Scheme 4 Synthesis of the 2-oxo-2,5-dihydro-1,2-oxaphospholes **12** by electrophilic cyclization of the phosphorylated β -hydroxyallenes **7** and **10**.

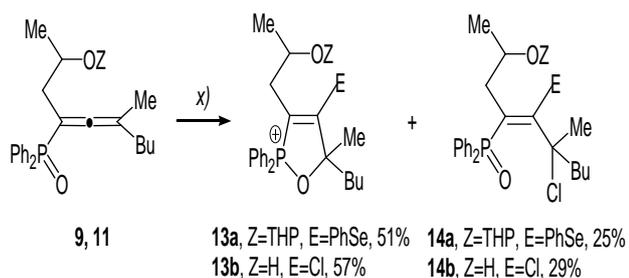
Reagents and Conditions: *ix*) Br_2 or PhSCl (1.2 eq), CH_2Cl_2 , -20°C , 2 h, rt, 2h, stirring, work-up, column chromatography.

In order to outline the general terms of this methodology, the reaction of the 1-(2-hydroxypropyl)-allenyl phosphine oxides with protected and unprotected hydroxyl group **9** and **11** with sulfonyl chloride or benzeneselenenyl chloride was thoroughly investigated.

Surprisingly, once we applied the current standard conditions to the 1,1-bifunctionalized allenes comprising a phosphine oxide and a 2-hydroxypropyl groups such as **9** and **11** (Scheme 5), the interaction afforded mixtures of the 2,2-diphenyl-2,5-dihydro-1,2-oxaphosphol-2-ium chlorides **13a,b** and the hept-(1*E*)-en-1-yl diphenyl phosphine oxides **14a,b** in the ratio about 2:1 in 76% and 86% total yield.

These reaction pathways may be interpreted as a result of the concurrent neighboring group participation of the phosphine oxide group as an internal nucleophile to give cyclic products **13a,b** and the highly regio- and stereoselective association of the external nucleophile, indicating a highly chemoselectively addition reaction of the electrophilic reagent to the $\text{C}^2\text{-C}^3$ -double bond of the allenic system with formation of the (1*E*)-2,3-adducts **14a,b**.

Christov, VC et al, *Bifunctionalized allenes. Part XVII. Synthesis of 2,5-Dihydro-1,2-Oxaphospholes and 4-Phosphoryl-3,6-Dihydro-2H-Pyrans by Electrophilic Cyclization and Coinage Metal-Catalyzed Cycloisomerization of Phosphorylated S-Hydroxyallenes*



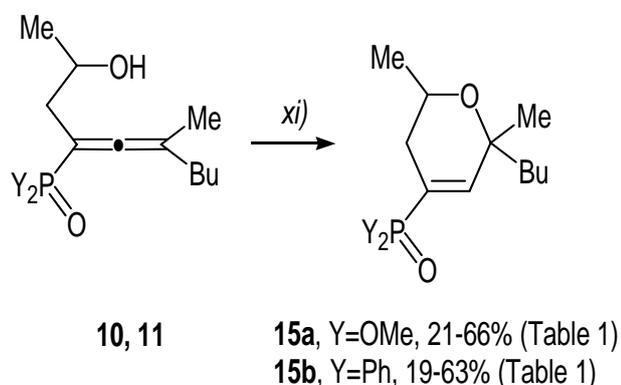
Scheme 5 Synthesis of the 2,5-dihydro-1,2-oxaphosphol-2-ium chlorides **13** and the 1-(2-hydroxypropyl)-hept-(1*E*)-en-1-yl phosphine oxides **14** by reaction of the phosphorylated β -hydroxyallenes **9** and **11** with electrophilic reagents.

Reagents and Conditions: x) PhSeCl or SO₂Cl₂ (1.2 eq), CH₂Cl₂, -20 °C, 2 h, rt, 2h, stirring, work-up, column chromatography.

The stereo selectivity could be explained by the favorable *trans* arrangement of the electrophile and the phosphine oxide group and *anti* attack of the external nucleophile. This is supposed to arise from attack on the allenic C²-C³ double bond *anti* to the phosphoryl group which assists in the cyclization by neighboring group participation as an internal nucleophile.

Coinage Metal-catalyzed Cycloisomerization of Phosphorylated S-Hydroxyallenes

In addition to the above mentioned preparation of 2,5-dihydro-1,2-oxaphospholes by electrophilic cyclization of the 1-(2-hydroxypropyl)-allenephosphonates **7** and **10** and allenyl phosphine oxides **9** and **11** due to the participation of the phosphonate or phosphine oxide neighboring group in the 5-*endo-trig* cyclization, the next step in our study was to explore the possibilities of the cycloisomerization reaction of the above mentioned compounds **10** and **11** in the presence of coinage metal salts as catalysts. We conducted the reaction under the optimized reaction conditions determined in the similar reactions of the phosphorylated α -hydroxyallenes earlier (Christov et al, 2015) – solvent methylene chloride, 5 mol % catalyst and room temperature. The reaction occurred via an 6-*endo-trig* cyclization to give the 4-phosphoryl-3,6-dihydro-2*H*-pyrans **15a,b** (Scheme 6).



Scheme 6 Synthesis of the 4-phosphoryl-3,6-dihydro-2*H*-pyrans **15** by coinage metal-catalyzed cycloisomerization of the phosphorylated β -hydroxyallenes **10** and **11**

Reagents and Conditions: xi) Coinage metal catalyst (Table 1, 5 mol %), CH₂Cl₂, rt, stirring (Table 1), work-up, column chromatography.

The type of catalyst and its influence on the yields and the reaction time of the cycloisomerization of the 1-(2-hydroxypropyl)-allenephosphonate **10** and allenyl phosphine oxide **11** was also of great interest to us. We performed a series of experiments with the sole intention of determining the best catalysts bearing in mind two criteria – highest yield and shortest reaction time (Scheme 6). We applied the following coinage metal salts as catalysts: AgNO₃, AgClO₄, AuCl, AuCl₃, Pd(PPh₃)₄, PtCl₂, CuCl₂, NiCl₂, SnCl₂, AlCl₃, PdCl₂, ZnCl₂, CuCl, CuBr, and CuI. The data reveal that both Au and Ag are excellent catalysts. Pd(I), Pt, and Cu(II) are very good catalysts for our experiments. It becomes obvious that Ni, Sn, Al, Pd(II), and Zn are relatively good catalysts. It is the Cu(I) catalysts that are bad. Table 1 presents the characteristics of all the above-mentioned catalysts in the cycloisomerization reaction of the compounds **10** and **11**.

Table 1 Synthesis of the 4-phosphoryl-3,6-dihydro-2*H*-pyrans **15** by coinage metal-catalyzed cycloisomerization of the phosphorylated β -hydroxyallenes **10** and **11**

Entry	Catalyst	Reaction time ^a 10 (h)	15a, Yield (%)	Reaction time ^a 11 (h)	15b, Yield (%)
1	AgNO ₃	15	60 ^b	13	59 ^c
2	AgClO ₄	12	60 ^c	11	58 ^c
3	AuCl	6	66 ^c	5	63 ^c
4	AuCl ₃	6	63 ^b	6	61 ^c
5	Pd(PPh ₃) ₄	16	51 ^c	13	49 ^b
6	PtCl ₂	18	54 ^b	16	48 ^c
7	CuCl ₂	17	50 ^c	17	52 ^b
8	NiCl ₂	24	44 ^c	21	38 ^b
9	SnCl ₂	24	39 ^b	25	32 ^c
10	AlCl ₃	24	40 ^b	22	40 ^c
11	PdCl ₂	20	39 ^b	17	35 ^c
12	ZnCl ₂	24	34 ^c	20	30 ^b
13	CuCl	42	26 ^c	37	24 ^b
14	CuBr	49	21 ^b	45	23 ^c
15	CuI	54	23 ^b	55	19 ^c

^a On the average; ^b Yields determined by ¹H- and ³¹P-NMR analysis; ^c Isolated yield by chromatographic purification on silica gel.

The results are explicit enough – a catalytic 6-*endo-trig* cycloisomerization occurs and the hydroxy group participates as an internal nucleophile to give the 4-phosphoryl-3,6-dihydro-2*H*-pyrans **15** in good yields.

CONCLUSIONS

In conclusion, a couple of new phosphorylated β -hydroxyallenes were synthesized by a convenient, efficient, atom economical and regioselective method. Reaction of the phosphorylated β -hydroxyallenes with protected or unprotected hydroxy groups with different electrophilic reagents occurs via 5-*endo-trig* cyclization. Treatment of the 1-(2-hydroxypropyl)-allenephosphonates with electrophiles gives the 2-methoxy-2-oxo-2,5-dihydro-1,2-oxaphospholes as a

result of the neighboring group participation of the phosphonate group in the cyclization, while the hept-1E-en-1-yl phosphine oxides were prepared as mixtures with the 2,5-dihydro-1,2-oxaphosphol-2-ium halides in a ratio of about 1:2 by chemo-, regio-, and stereo-selective electrophilic addition to the C²-C³-double bond in the allene moiety and subsequent concurrent attack of the external (chloride anion) and internal (phosphine oxide group) nucleophiles. We have developed a coinage metal-catalyzed cycloisomerization reaction of the phosphorylated α -hydroxyallenes, which provides an efficient route to the 4-phosphoryl-3,6-dihydro-2H-pyrans which are produced as a result of the participation of the neighboring hydroxy group as an internal nucleophile in the cyclization process.

Due to the easy availability of starting materials, the convenient operation and mild conditions, the readily availability of the reagents and catalysts, the good yields and the usefulness of the 2,5-dihydro-1,2-oxaphospholes and the 4-phosphoryl-3,6-dihydro-2H-pyrans, the cyclization reactions may show potential and will be useful in organic synthesis as well in their application in target-oriented synthesis. Further investigation on the chemistry of other phosphorylated allenols for the synthesis of different heterocyclic systems is being intensively carried out in our laboratory. Moreover, results of an initial investigation of the biological activity of the compounds prepared were encouraging, and the antibacterial and antifungal activities of selected compounds are now under investigation in our University.

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